



QUALITATIVE AND QUANTITATIVE PROFILES OF MAJOR MACRONUTRIENT ELEMENTS PRESENT IN ENERGY DRINKS POPULAR IN INDIA USING ICP-MS

Pooja Rana^{1*}, Arvind Choudhary², Munish Kumar Mishra³, Lav Kesharwani⁴, Suchit A. John⁵, Poonam Prakash⁶

¹*PhD Research scholar, Department of Forensic Science, Sam Higginbottom University of Agriculture, Technology and Sciences, Prayagraj, Uttar Pradesh, pooja.rana.19.10@gmail.com

²Senior Analyst, Rohilkhand Laboratory and Research Center, Bareilly, Uttar Pradesh, choudharyarvind2027@gmail.com

^{3,4}Professor, Department of Forensic Science, Sam Higginbottom University of Agriculture, Technology and Sciences, Prayagraj, Uttar Pradesh, dr.munishm@gmail.com, lavkesharwani@gmail.com

⁵Professor, Head, Department of Forensic Science, Sam Higginbottom University of Agriculture, Technology and Sciences, Prayagraj, Uttar Pradesh, suchit.john@shiats.edu.in

⁶Professor, Department of Chemistry, Sam Higginbottom University of Agriculture, Technology and Sciences, Prayagraj, Uttar Pradesh, poonam.prakash@shiats.edu.in

***Corresponding author-** Pooja Rana

*Department of Forensic Science, Sam Higginbottom University of Agriculture, Technology and Sciences, Prayagraj, Uttar Pradesh- 211007. Email: pooja.rana.19.10@gmail.com

Abstract

Energy drinks market is increasingly popular in India, where they have been marketed with claims to provide instant boost of energy. They have been equally popular subject of study by food scientists as well where major contents like caffeine, sugars, etc. present in them have been extensively studied by the researchers but less focus has been paid on their elemental composition. The elemental profiling of these drinks becomes a subject of interest also because the consumers consistently consuming these drinks are more susceptible to be negatively impacted by the presence of certain elemental components present in these products, especially because of their high concentrations. Present study investigates the qualitative and quantitative profiles of macro nutrient elements present in energy drinks from ten popular brands in India. Macronutrient elements are required in large amounts for normal and vital functioning of physiological processes in body. In this study, quantitative assessment of 5 such macronutrient elements (Sodium (Na), Magnesium (Mg), Aluminum (Al), Potassium (K), Calcium (Ca)) in energy drink samples of 10 different brands was carried out using microwave assisted acid decomposition and quantitation through ICP-MS instrument which is capable of analyzing several samples rapidly and permitted the identification of multiple metallic elements present in those samples which may be useful for investigations of food frauds in various legal issues. The findings revealed several brands exceeded the permissible limits set by the Food Safety and Standards Authority of India (FSSAI) for key elements, raising significant health concerns.

Key words: *Macronutrient, ICP-MS, energy drinks, elemental profiling, FSSAI*

1. Introduction

Energy drinks are primarily composed of carbohydrates, vitamins, and a variety of minerals, serving as a source of bodily energy¹. These drinks have rapidly gained global popularity as functional beverages designed to enhance physical and cognitive performance through the inclusion of stimulants, vitamins, and other bioactive compounds. Initially targeting athletes, these drinks have expanded their market to encompass adolescents, young adults, and even the elderly, driven by aggressive marketing and changing consumer lifestyles. While they are associated with increased energy and endurance, their diverse formulations and high stimulant content raise concerns about potential health risks.

Most studies on energy drinks focus on caffeine content, limited research has examined their metal composition. Metallic elements, although essential for enzyme metabolism and maintaining physical health, can become detrimental when consumed in excess, especially for individuals who habitually consume these beverages². The analysis of various elements in energy drinks serves multiple purposes, primarily focusing on their nutritional significance and potential toxicity. Aluminum (Al) is monitored due to its toxicity, even in trace amounts. Essential minerals such as calcium (Ca), magnesium (Mg), potassium (K), and sodium (Na) are assessed for their nutritional contributions, with their concentrations varying from major to minor or trace levels³. Calcium plays a crucial role in bone health, while magnesium is vital for enzymatic functions. Potassium and sodium are key electrolytes involved in maintaining fluid balance and nerve function³. Understanding the presence and concentration of these elements is essential for evaluating the health implications of energy drink consumption. The analysis of metal content in food samples is highly significant as it offers valuable insights into their nutritional composition, facilitates the development of food composition databases, and ensures the monitoring of antinutritional elements or potentially toxic substances⁴. The packaging material used for energy drinks, including polyethylene terephthalate (PET) and glass bottles, as well as aluminum and steel cans, plays a significant role in influencing the final product's characteristics⁵. Metal leaching during manufacturing can introduce impurities⁶, and the levels of metals in energy drinks are influenced by raw materials, production processes, and additives⁵.

Analytical techniques such as flame atomic absorption spectrometry (FAAS)^{7,8}, electrothermal atomic absorption spectrometry⁹, microwave plasma atomic emission spectrometry¹⁰, inductively coupled plasma optical emission spectrometry (ICP-OES)^{11,12} and inductively coupled plasma mass spectrometry (ICP-MS)¹³ have been employed to quantify metallic elements in food and beverages. In this study, the concentrations of aluminum (Al), calcium (Ca), magnesium (Mg), sodium (Na), and potassium (K) were determined in 100 energy drink samples from 10 popular Indian brands, packaged in aluminum cans and PET bottles, using ICP-MS.

2. Methodology

2.1. Instrumentation

An ultrasonic bath was used for the degassing of soft drink samples. For sample preparation, after degassing, samples were poured in Teflon® lined digestion vessels, placed in microwave digester (Multiwave GO Plus, Anton Paar) and using multi-step programming of temperature-time program, they were subjected to high temperature (120°C, 150°C) and pressure (300 psi) for digestion of organic contents in the sample to reduce matrix interference during sample analysis by ICP-MS. Samples were then transferred to Tarson tubes and diluted to a known volume using ultrapure Milli-Q water, which were then loaded in tray and subjected to ICP-MS instrument.

Inductively coupled plasma mass spectrometer (ICP MS) (Agilent ICP-MS (7800 series)) was used to quantify macronutrient elements (Na, Mg, Al, K and Ca) in energy drinks samples. The determination of macro and micronutrients were carried out under the conditions shown in Table 1 below:

Table 1: Set-up conditions of ICP-MS instrument

Parameter	Condition
RF Power (W)	1550
Carrier gas (l/min)	1.029
Plasma gas flow (l/min)	15
Auxiliary gas flow (l/min)	1.0
Spray chamber	Water cooled double pass
Sample chamber temperature (°C)	2
Lens voltage (V)	8.2
Mass resolution	0.8
Integration time points/ms	3
Points per peak	3
Replicates	3
Carrier gases	
Argon gas (high purity 99.99%) (kPa)	630
Helium gas (ultra high purity- 99.999%) (kPa)	120

The concentration of the analytes of interest was automatically calculated by the software of the ICP-MS instrument. The following steps were performed for each element: The count rates were corrected according to the correction functions chosen. For the count rates measured in the zero members, calibration and test solutions were normalized on the count rates of the internal standard. The calibration function was calculated. By the use of the count rates, the calibration function and the dilution factor the concentrations of the elements were calculated.

For analytical quality control, blank solutions and reference samples of comparable matrix having reliably known contents of the elements to be determined were analyzed in parallel with all the series of samples analyzed. The reference samples were subjected to all the steps in the method, starting from the digestion.

2.2. Reagents and solutions

All reagents and reference materials used in this study were of ICP-MS/Trace metal grade and all the lab wares and instruments used were sterile. The procedure for determination of total acid-extractable concentrations of aluminum (Al), calcium (Ca), magnesium (Mg), sodium (Na), and potassium (K) in energy drinks using microwave assisted acid decomposition ICP-MS was followed^{14,15}.

Firstly, standard stock solution of 1000 ppm concentration was prepared for Al, Ca, Mg, Na, K and Hg. These standard solutions were diluted using 2% nitric acid (ultra pure) at different concentrations (0.2 ppm, 0.05 ppm, 0.02 ppm, 0.005 ppm, 0.001 ppm, 0.0005 ppm, 0.0001 ppm- all elements except Hg and 0.02 ppm, 0.005 ppm, 0.002 ppm, 0.0005 ppm, 0.0001 ppm, 0.00005 ppm, 0.00001 ppm - Hg).

2.3. Sample collection

Energy drinks (100 samples; 10 samples each from 10 different brands) contained in PET and glass bottles were procured from various online and offline stores in Prayagraj, Uttar Pradesh at different time period and stored at 4°C in sterile conditions until digestion. The samples were tested at Instrumental Laboratory, Rohilkhand Laboratory and Research Center, Bareilly, Uttar Pradesh in 2022.

3. Results and discussion

Tables 2-11 show the concentrations of the five elements determined by ICP MS in energy drink samples from brands B1-B10. These elements were present in high concentrations (Na, Mg, K, Ca and Al). These concentration levels were expected, because the first four elements are considered

element macronutrients, or are present in high concentrations in food and are needed in extremely large amounts for the metabolism¹⁶, whereas, Al turns out to be undesirable micronutrient element which was also present in exceptionally high amounts in energy drinks stored in metal cans. The concentrations of all the analytes of choice for this study were present in high concentrations, as opposed to the permissible levels set by Food Safety and Standards Authority of India (FSSAI).

Table 2: Concentration of macronutrient elements in samples from brand B1

Permissible value as per FSSAI		200 ppm	50 ppm	30 ppb	--	20-75 ppm
Mfg. Month	Sample	Na [He] ppm	Mg [He] ppm	Al [He] ppb	K [He] ppm	Ca [He] ppm
Feb, 2022	E1	481.88	92.38	172.03	360.43	293.70
Feb, 2022	E2	231.34	87.01	160.66	224.31	104.61
Apr, 2022	E21	369.19	80.62	157.69	250.18	122.16
Apr, 2022	E22	147.69	65.01	148.67	141.07	78.25
Jun, 2022	E41	256.49	62.04	75.65	128.03	80.74
Jun, 2022	E42	78.09	30.03	71.82	96.10	61.58
Aug, 2022	E61	168.04	57.05	69.18	89.07	51.95
Aug, 2022	E62	73.48	23.03	53.03	80.03	50.32
Oct, 2022	E81	57.97	20.20	23.01	40.03	40.56
Oct, 2022	E82	58.68	14.90	22.41	41.19	39.22

Table 3: Concentration of macronutrient elements in samples from brand B2

Permissible value as per FSSAI		200 ppm	50 ppm	30 ppb	--	20-75 ppm
Mfg. Month	Sample	Na [He] ppm	Mg [He] ppm	Al [He] ppb	K [He] ppm	Ca [He] ppm
Feb, 2022	E3	498.99	136.78	155.60	490.10	308.64
Feb, 2022	E4	303.30	92.46	153.86	358.03	170.09
Apr, 2022	E23	330.07	114.98	93.79	381.09	270.47
Apr, 2022	E24	154.67	85.98	92.71	290.08	129.35
Jun, 2022	E43	249.84	106.06	88.90	210.36	202.06
Jun, 2022	E44	133.15	49.36	81.61	150.05	84.34
Aug, 2022	E63	193.08	85.70	42.71	190.81	181.43
Aug, 2022	E64	98.12	25.84	39.36	96.26	80.95
Oct, 2022	E83	33.32	15.25	13.58	104.01	47.47
Oct, 2022	E84	38.00	13.60	11.29	91.07	43.00

Table 4: Concentration of macronutrient elements in samples from brand B3

Permissible value as per FSSAI		200 ppm	50 ppm	30 ppb	--	20-75 ppm
Mfg. Month	Sample	Na [He] ppm	Mg [He] ppm	Al [He] ppb	K [He] ppm	Ca [He] ppm
Feb, 2022	E5	547.52	200.05	205.66	398.64	289.17
Feb, 2022	E6	414.91	118.60	156.68	351.08	240.17
Apr, 2022	E25	482.01	160.04	157.49	311.58	231.02
Apr, 2022	E26	371.51	89.90	120.01	290.65	176.41
Jun, 2022	E45	340.18	95.89	85.07	250.39	183.67
Jun, 2022	E46	268.35	51.83	69.29	211.50	120.84
Aug, 2022	E65	232.47	63.26	49.30	185.26	105.94

Aug, 2022	E66	187.51	41.08	41.60	141.37	83.02
Oct, 2022	E85	141.22	30.07	29.96	78.44	61.17
Oct, 2022	E86	136.08	28.06	27.44	76.38	58.19

Table 5: Concentration of macronutrient elements in samples from brand B4

Permissible value as per FSSAI		200 ppm	50 ppm	30 ppb	--	20-75 ppm
Mfg. Month	Sample	Na [He] ppm	Mg [He] ppm	Al [He] ppb	K [He] ppm	Ca [He] ppm
Feb, 2022	E7	477.20	257.82	556.06	514.76	679.16
Feb, 2022	E8	391.07	218.20	505.10	493.06	512.16
Apr, 2022	E27	380.85	190.16	549.86	460.15	481.19
Apr, 2022	E28	304.46	154.64	500.90	341.01	457.05
Jun, 2022	E47	310.84	114.01	472.89	375.04	348.01
Jun, 2022	E48	256.00	91.01	454.62	260.40	309.16
Aug, 2022	E67	190.49	60.16	288.00	198.17	271.15
Aug, 2022	E68	156.42	40.81	210.03	157.01	225.02
Oct, 2022	E87	113.61	43.26	91.05	78.09	90.08
Oct, 2022	E88	80.32	39.53	58.90	69.37	85.41

Table 6: Concentration of macronutrient elements in samples from brand B5

Permissible value as per FSSAI		200 ppm	50 ppm	30 ppb	--	20-75 ppm
Mfg. Month	Sample	Na [He] ppm	Mg [He] ppm	Al [He] ppb	K [He] ppm	Ca [He] ppm
Feb, 2022	E9	408.52	186.20	130.63	539.43	335.73
Feb, 2022	E10	356.01	157.08	97.31	470.02	220.46
Apr, 2022	E29	362.59	134.00	104.75	425.04	299.06
Apr, 2022	E30	301.00	104.62	87.09	301.76	140.21
Jun, 2022	E49	323.74	86.61	62.42	259.25	182.13
Jun, 2022	E50	272.16	64.22	56.09	223.85	100.01
Aug, 2022	E69	134.01	43.27	46.11	164.22	108.06
Aug, 2022	E70	103.24	32.25	40.47	114.07	67.44
Oct, 2022	E89	95.41	20.04	20.39	90.06	51.12
Oct, 2022	E90	91.48	16.81	20.29	87.28	48.93

Table 7: Concentration of macronutrient elements in samples from brand B6

Permissible value as per FSSAI		200 ppm	50 ppm	30 ppb	--	20-75 ppm
Mfg. Month	Sample	Na [He] ppm	Mg [He] ppm	Al [He] ppb	K [He] ppm	Ca [He] ppm
Feb, 2022	E11	409.058	119.204	9.877	419.927	247.119
Feb, 2022	E12	370.332	83.467	9.254	383.149	112.049
Apr, 2022	E31	320.164	86.156	11.951	342.561	189.666
Apr, 2022	E32	281.836	65.155	11.473	204.538	98.798
Jun, 2022	E51	212.096	49.076	8.683	254.532	103.308
Jun, 2022	E52	161.819	41.023	8.287	138.486	80.429
Aug, 2022	E71	154.359	42.069	7.891	194.459	79.457
Aug, 2022	E72	110.333	36.067	7.852	116.443	64.175
Oct, 2022	E91	62.233	23.065	8.394	87.121	61.362
Oct, 2022	E92	62.069	20.063	6.426	85.120	57.214

Table 8: Concentration of macronutrient elements in samples from brand B7

Permissible value as per FSSAI		200 ppm	50 ppm	30 ppb	--	20-75 ppm
Mfg. Month	Sample	Na [He] ppm	Mg [He] ppm	Al [He] ppb	K [He] ppm	Ca [He] ppm
Feb, 2022	E13	357.19	146.04	29.94	433.00	407.34
Feb, 2022	E14	217.47	83.52	21.32	266.21	188.66
Apr, 2022	E33	221.88	90.15	6.25	284.12	258.33
Apr, 2022	E34	185.09	51.04	8.41	197.48	139.97
Jun, 2022	E53	129.00	58.40	11.09	151.06	114.40
Jun, 2022	E54	90.80	42.02	8.36	103.25	94.36
Aug, 2022	E73	67.36	43.44	10.23	84.13	71.33
Aug, 2022	E74	49.48	35.00	11.20	52.37	60.14
Oct, 2022	E93	23.30	20.43	19.72	30.20	31.34
Oct, 2022	E94	22.03	19.74	19.31	29.36	31.05

Table 9: Concentration of macronutrient elements in samples from brand B8

Permissible value as per FSSAI		200 ppm	50 ppm	30 ppb	--	20-75 ppm
Mfg. Month	Sample	Na [He] ppm	Mg [He] ppm	Al [He] ppb	K [He] ppm	Ca [He] ppm
Feb, 2022	E15	548.99	175.11	95.51	488.09	318.64
Feb, 2022	E16	333.00	117.41	73.06	338.02	188.01
Apr, 2022	E35	431.01	134.98	83.73	374.08	276.42
Apr, 2022	E36	254.13	103.24	62.71	297.15	132.05
Jun, 2022	E55	349.12	108.08	68.08	198.38	212.06
Jun, 2022	E56	147.30	89.36	41.65	140.21	104.24
Aug, 2022	E75	273.02	85.74	52.74	171.52	154.00
Aug, 2022	E76	138.11	45.84	34.36	106.23	75.40
Oct, 2022	E95	114.02	28.09	23.52	124.02	78.47
Oct, 2022	E96	89.32	25.25	21.01	96.36	71.05

Table 10: Concentration of macronutrient elements in samples from brand B9

Permissible value as per FSSAI		200 ppm	50 ppm	30 ppb	--	20-75 ppm
Mfg. Month	Sample	Na [He] ppm	Mg [He] ppm	Al [He] ppb	K [He] ppm	Ca [He] ppm
Feb, 2022	E17	489.36	109.81	103.70	368.00	127.03
Feb, 2022	E18	311.01	90.34	82.65	310.00	87.36
Apr, 2022	E37	399.01	114.00	98.81	300.83	102.66
Apr, 2022	E38	287.62	74.62	73.11	216.67	81.21
Jun, 2022	E57	371.06	106.61	92.71	315.07	91.92
Jun, 2022	E58	200.01	52.09	59.42	204.85	76.43
Aug, 2022	E77	259.32	83.27	63.33	269.35	85.35
Aug, 2022	E78	177.60	51.81	21.29	188.63	68.63
Oct, 2022	E97	186.20	60.04	9.69	176.26	34.00
Oct, 2022	E98	178.03	48.62	10.80	152.85	31.60

Table 11: Concentration of macronutrient elements in samples from brand B10

Permissible value as per FSSAI		200 ppm	50 ppm	30 ppb	--	20-75 ppm
Mfg. Month	Sample	Na [He] ppm	Mg [He] ppm	Al [He] ppb	K [He] ppm	Ca [He] ppm
Feb, 2022	E19	493.01	119.20	98.52	511.46	355.46
Feb, 2022	E20	351.09	99.51	87.35	357.49	304.23
Apr, 2022	E39	332.60	90.07	101.07	483.54	271.35
Apr, 2022	E40	302.06	81.03	61.43	280.12	260.36
Jun, 2022	E59	292.01	113.47	72.00	397.93	247.12
Jun, 2022	E60	256.03	64.01	40.54	136.15	192.05
Aug, 2022	E79	231.84	73.15	32.19	91.53	186.31
Aug, 2022	E80	187.02	52.07	33.58	78.56	130.67
Oct, 2022	E99	190.36	67.17	19.20	63.44	79.18
Oct, 2022	E100	122.90	48.90	18.92	59.02	74.03

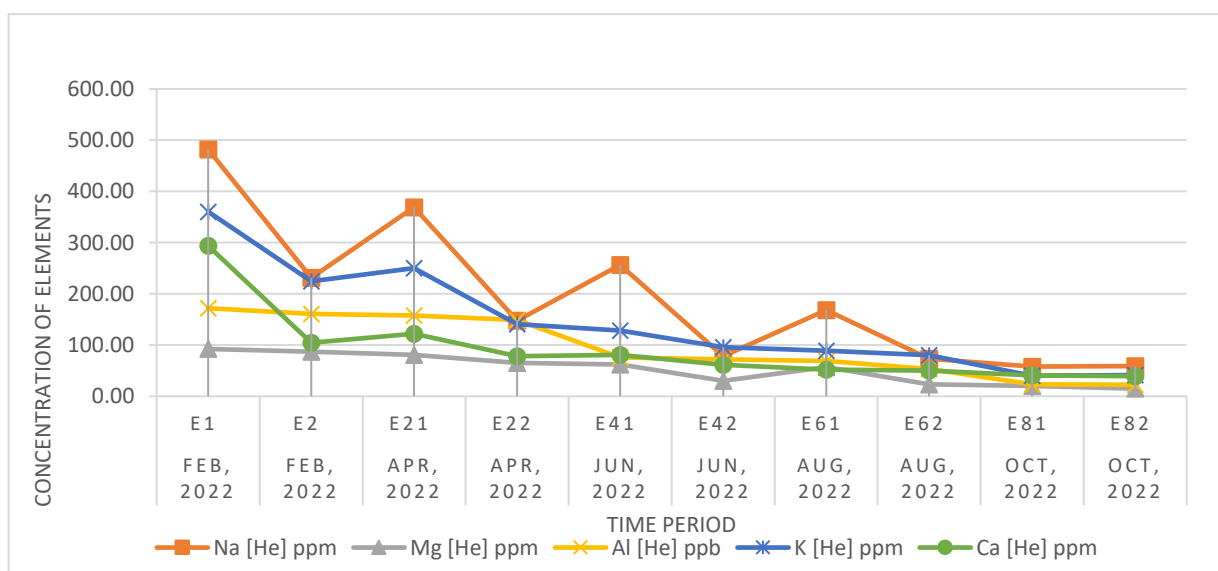


Figure 1: Trend in concentration of macronutrient elements in samples from brand B1

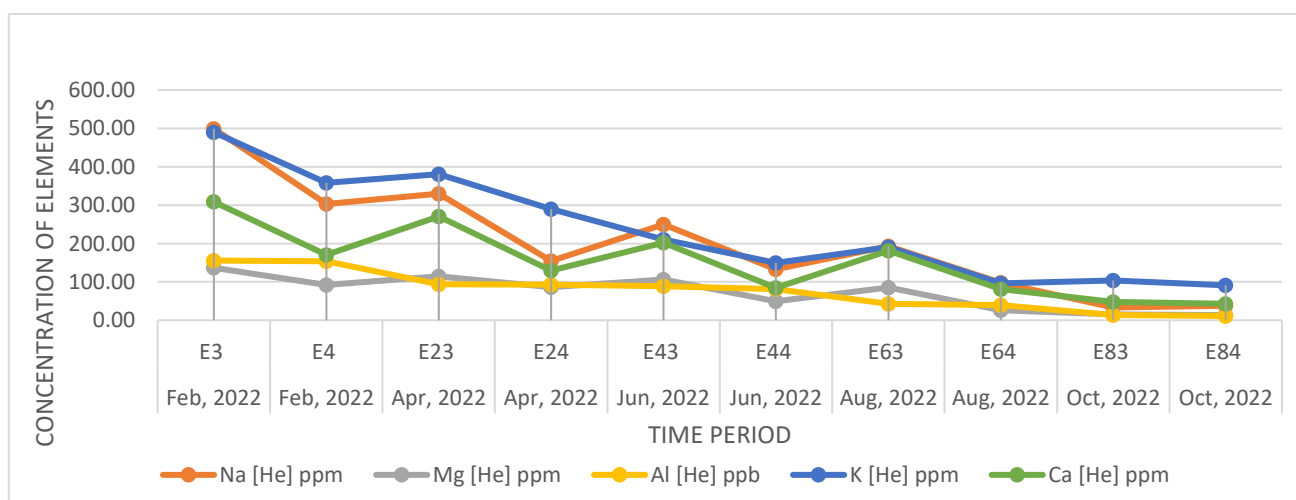


Figure 2: Trend in concentration of macronutrient elements in samples from brand B2

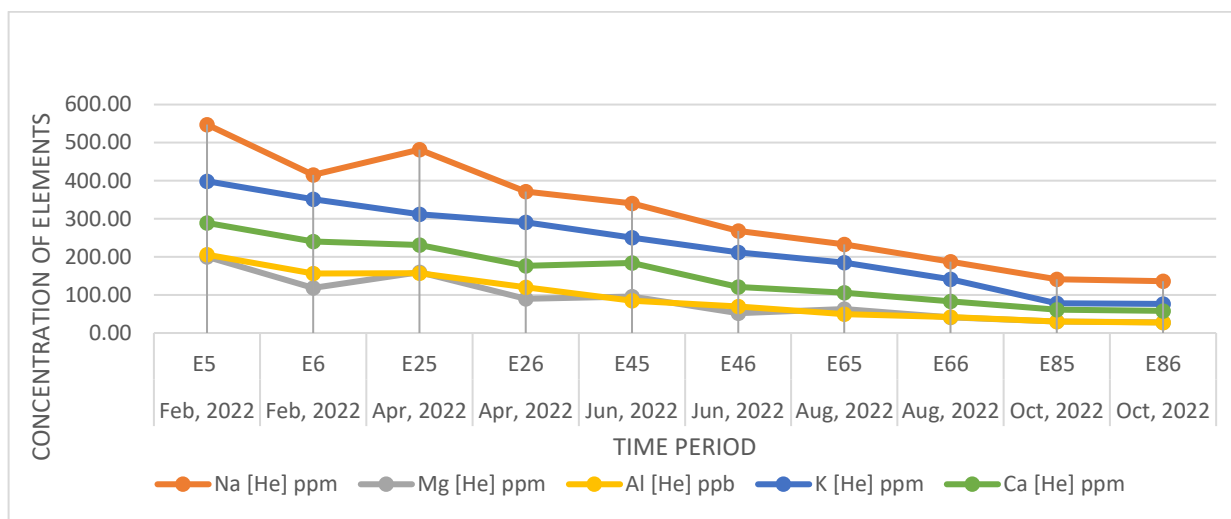


Figure 3: Trend in concentration of macronutrient elements in samples from brand B3

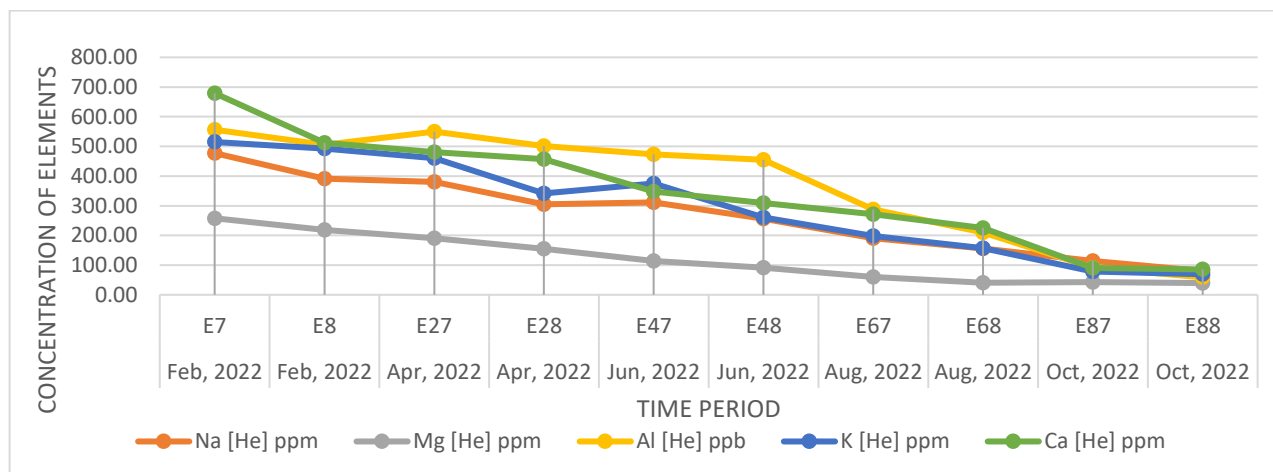


Figure 4: Trend in concentration of macronutrient elements in samples from brand B4

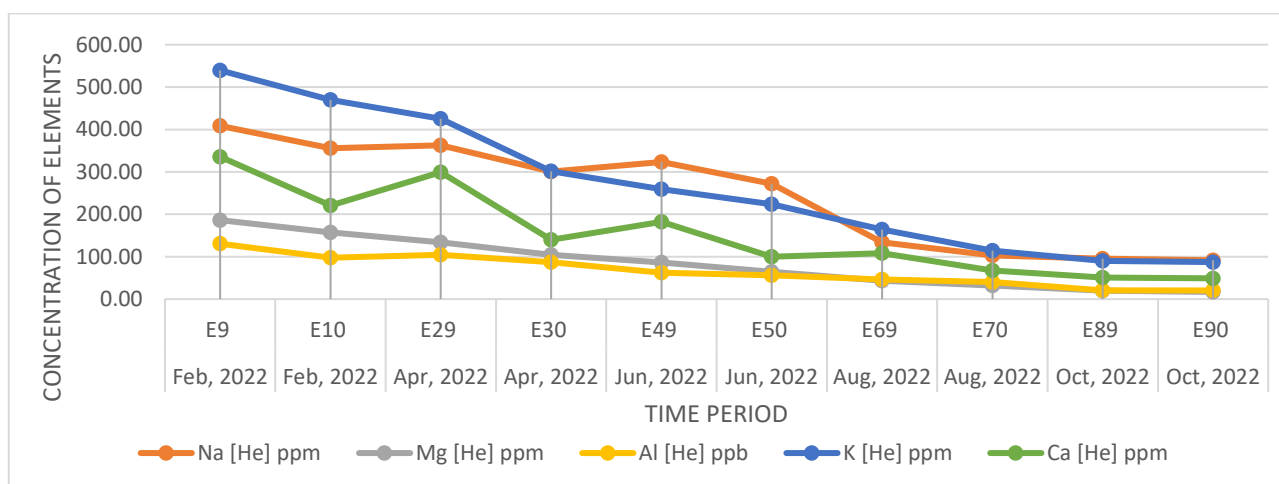


Figure 5: Trend in concentration of macronutrient elements in samples from brand B5

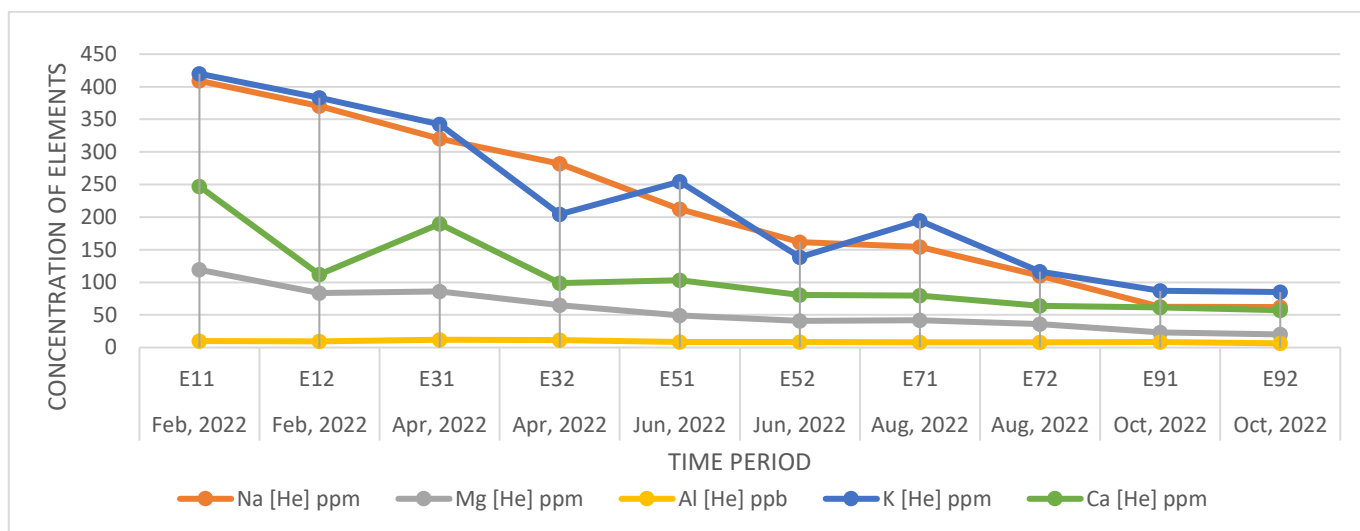


Figure 6: Trend in concentration of macronutrient elements in samples from brand B6

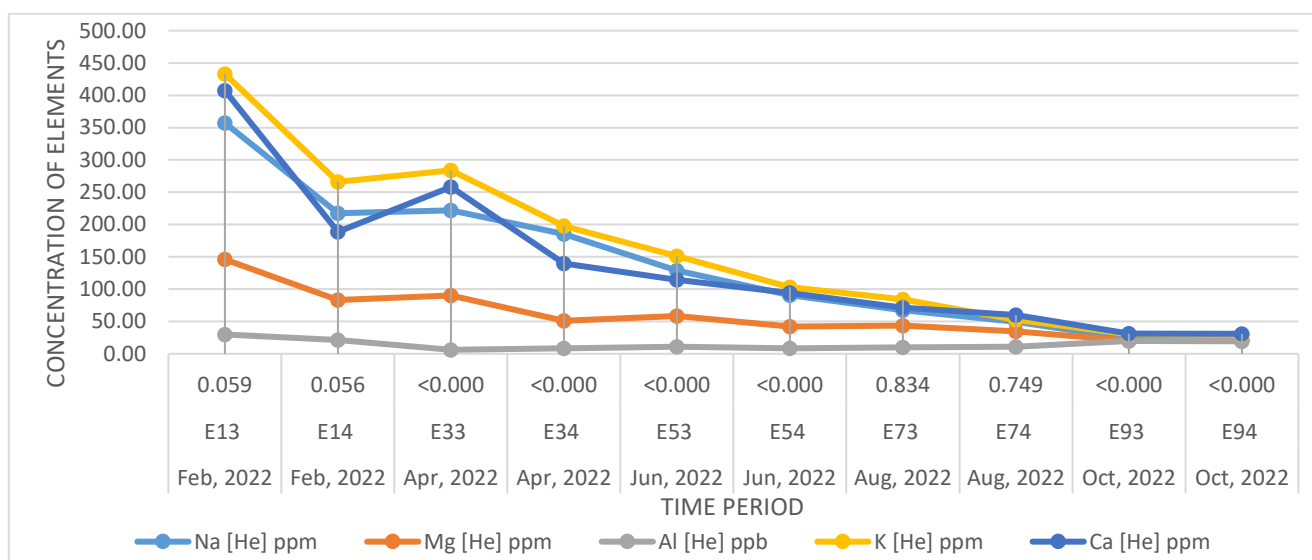


Figure 7: Trend in concentration of macronutrient elements in samples from brand B7

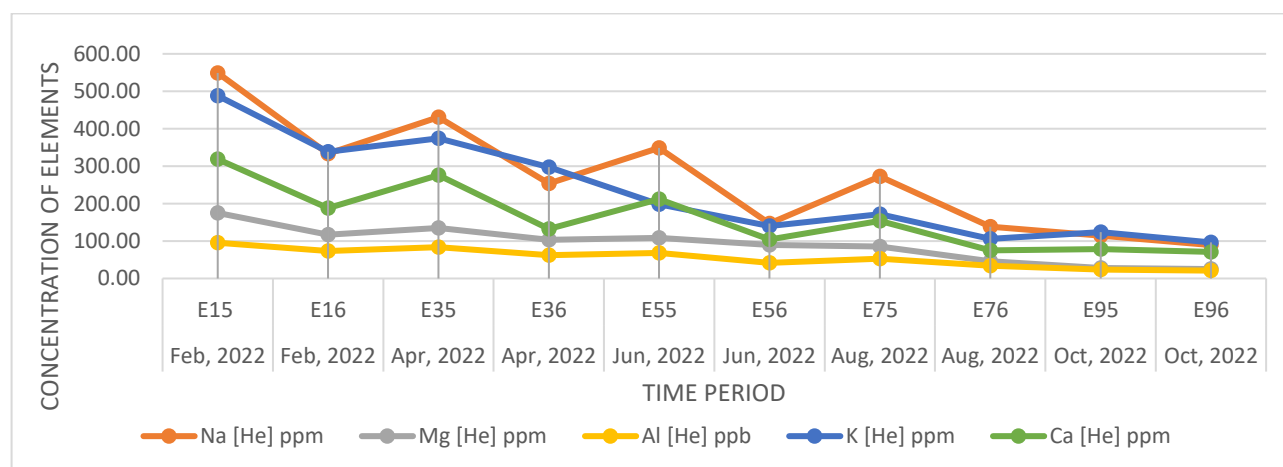


Figure 8: Trend in concentration of macronutrient elements in samples from brand B8

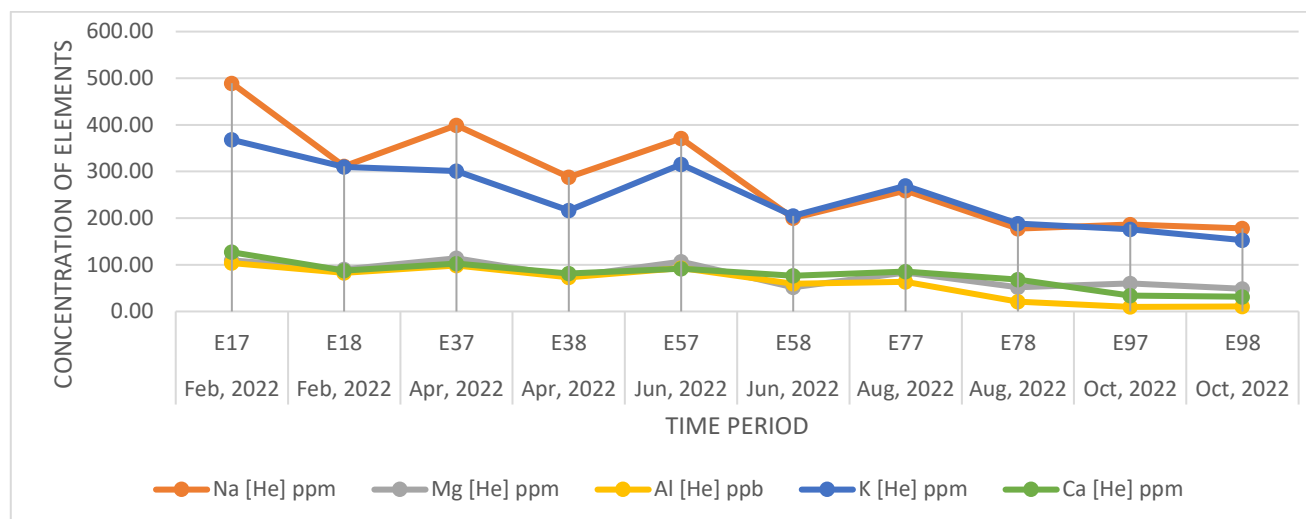


Figure 9: Trend in concentration of macronutrient elements in samples from brand B9

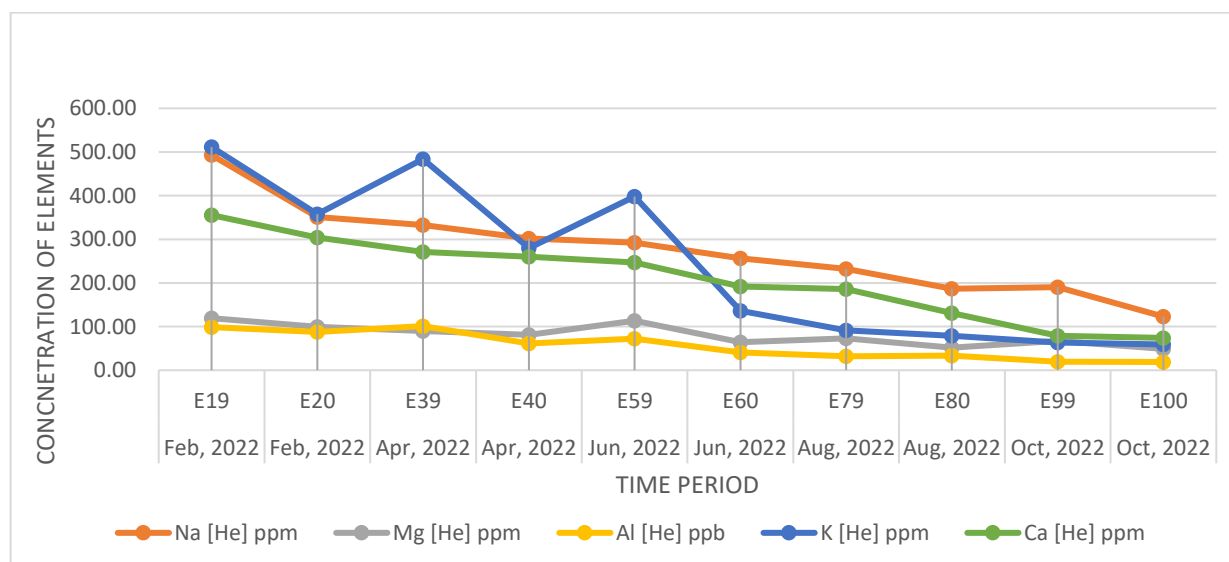


Figure 10: Trend in concentration of macronutrient elements in samples from brand B10

The analysis of macronutrient concentrations in energy drink samples from ten different brands revealed significant variations in sodium (Na), magnesium (Mg), aluminum (Al), potassium (K), and calcium (Ca) levels. Sodium concentrations across brands ranged from a maximum of 548.99 ppm (E15, B8) to a minimum of 22.03 ppm (E94, B7). Brand B8 exhibited the highest sodium concentration, whereas brand B7 showed the lowest. All brands had at least one sample exceeding the permissible FSSAI limit of 200 ppm, indicating non-compliance.

Magnesium levels varied between 257.82 ppm (E7, B4) and 13.60 ppm (E84, B2). The highest concentrations were observed in brand B4, while the lowest was found in brand B2. Many samples from multiple brands exceeded the FSSAI limit of 50 ppm, highlighting non-compliance.

Aluminum content spanned from 556.06 ppb (E7, B4) to 6.25 ppb (E33, B7). Brand B4 recorded the highest aluminum concentration, far exceeding the permissible limit of 30 ppb, whereas B7 had the lowest. Most samples in brands B4 and B5 significantly exceeded the limit, indicating a potential health risk. Another interesting trend which was seen was that the samples which were packaged in PET bottles (brands B6 and B7) had Al concentrations well within the permissible limits set by FSSAI whereas samples from other brands that were packaged in aluminum cans showed significantly high concentrations of Al in their samples.

Potassium levels ranged from 539.43 ppm (E9, B5) to 29.36 ppm (E94, B7). The highest concentration was found in brand B5, while B7 had the lowest levels. Since no FSSAI standard is specified for potassium, compliance was assessed only against the product label.

Calcium concentrations were highest at 679.16 ppm (E7, B4) and lowest at 31.05 ppm (E94, B7). Brand B4 showed excessive calcium levels, surpassing the permissible range of 20-75 ppm in most of its samples, while brand B7 had the lowest concentration. Several other brands, including B1, B2, and B3, also exhibited samples exceeding the upper FSSAI limit, indicating non-compliance.

Previous research^{17,18} has also provided valuable context for this study which highlight the influence of packaging materials on the elemental content of energy drinks, specifically aluminum cans. These studies draw attention to the potential leaching of elements such as aluminum into the beverage, which is critical when interpreting the findings of this study. Similarly, another research in Poland¹⁹ observed elevated levels of calcium and magnesium in energy drinks, which parallels some of the findings in this research.

Overall, sodium, magnesium, aluminum, and calcium levels frequently surpassed the permissible limits set by FSSAI, indicating inconsistencies in formulation across brands. These deviations from regulatory standards highlight potential health concerns and emphasize the need for strict quality control measures in the energy drink industry.

4. Conclusion

The present study on energy drinks 10 popular brands in India has revealed significant variations across different brands. The study found that Na, Mg, Al, K and Ca levels frequently exceeded FSSAI regulatory limits, raising concerns about compliance and potential health risks. Na, Mg, Ca and K concentrations in several brands surpassed the permissible limit by FSSAI suggesting inconsistencies in formulation and non-compliance with FSSAI standards. Al contamination was notably higher in brands using aluminum cans for packaging, reinforcing concerns regarding elemental leaching from container materials.

These findings emphasize the need for stringent regulatory lapse and quality control measures in the energy drink industry to ensure consumer safety. The observed variations, coupled with similarities to previous studies across the world, highlight the broader implications of packaging materials and ingredient inconsistencies. Future research should focus on further investigating the impact of packaging on elemental contamination, as well as potential health effects associated with excessive consumption of these macronutrients. By addressing these concerns, regulatory authorities and manufacturers can work toward ensuring safer and more standardized formulations for energy drinks in the Indian market.

References

1. Bulut, B., Beyhun, N. E., Topbaş, M., & Can, G. (2014). Energy drink use in university students and associated factors. *Journal of community health*, 39, 1004-1011.
2. Ribeiro, P., Morais, T. B. D., Colugnati, F. A. B., & Sigulem, D. M. (2003). Tabelas de composição química de alimentos: análise comparativa com resultados laboratoriais. *Revista de Saúde Pública*, 37, 216-225.
3. Otles, S. (Ed.). (2011). *Methods of analysis of food components and additives*. CRC press.
4. Mandlate, J. S., Soares, B. M., Seeger, T. S., Dalla Vecchia, P., Mello, P. A., Flores, E. M., and Duarte, F. A. (2017). Determination of cadmium and lead at sub-ppt level in soft drinks: An efficient combination between dispersive liquid-liquid microextraction and graphite furnace atomic absorption spectrometry. *Food Chemistry*, 221, 907-912.
5. Siegmund, B., Derler, K., & Pfannhauser, W. (2004). Chemical and sensory effects of glass and laminated carton packages on fruit juice products—still a controversial topic. *LWT-Food Science and Technology*, 37(4), 481-488.
6. Jellesen, M. S., Rasmussen, A. A., and Hilbert, L. R. (2006). A review of metal release in the food industry. *Materials and Corrosion*, 57(5), 387–393.

7. Sweileh, J. A., Misef, K. Y., El-Sheikh, A. H., & Sunjuk, M. S. (2014). Development of a new method for determination of aluminum (Al) in Jordanian foods and drinks: Solid phase extraction and adsorption of Al³⁺-D-mannitol on carbon nanotubes. *Journal of Food Composition and Analysis*, 33(1), 6-13.
8. Schiavo, D., Neira, J. Y., & Nóbrega, J. A. (2008). Direct determination of Cd, Cu and Pb in wines and grape juices by thermospray flame furnace atomic absorption spectrometry. *Talanta*, 76(5), 1113-1118.
9. Francisco, B. B. A., Brum, D. M., & Cassella, R. J. (2015). Determination of metals in soft drinks packed in different materials by ETAAS. *Food chemistry*, 185, 488-494.
10. Ozbek, N., & Akman, S. (2015). Determination of boron in Turkish wines by microwave plasma atomic emission spectrometry. *LWT-Food Science and Technology*, 61(2), 532-535.
11. Froes, R. E., Neto, W. B., Naveira, R. L., Silva, N. C., Nascentes, C. C., & da Silva, J. B. (2009). Exploratory analysis and inductively coupled plasma optical emission spectrometry (ICP OES) applied in the determination of metals in soft drinks. *Microchemical Journal*, 92(1), 68-72.
12. Leśniewicz, A., Grzesiak, M., Żyrnicki, W., & Borkowska-Burnecka, J. (2016). Mineral composition and nutritive value of isotonic and energy drinks. *Biological trace element research*, 170, 485-495.
13. Kılıç, S., Yenisoy-Karakaş, S., & Kılıç, M. (2015). Metal contamination in fruit juices in Turkey: method validation and uncertainty budget. *Food analytical methods*, 8, 2487-2495.
14. British Standard, BS EN 15763: 2009 Foodstuffs- Determination of trace elements- Determination of arsenic, cadmium, mercury and lead in foodstuffs by inductively coupled plasma mass spectrometry (ICPMS) after pressure digestion.
15. U S Food and Drug Administration , Elemental Analysis Manual, Inductively Coupled Plasma-Mass Spectrometric Determination of Arsenic, Cadmium, Chromium, Lead, Mercury, and Other Elements in Food Using Microwave Assisted Digestion. Version 1.1 (March 2015)
16. dos Santos Silva, E., da Silva, E. G. P., dos Santos Silva, D., Novaes, C. G., Amorim, F. A. C., Dos Santos, M. J. S., & Bezerra, M. A. (2019). Evaluation of macro and micronutrient elements content from soft drinks using principal component analysis and Kohonen self-organizing maps. *Food chemistry*, 273, 9-14.
17. SA, B., a Braskem, C., & Industriais, C. (2004). *Descritivo de processo*. São Paulo, Brazil.
18. Francisco, B. B. A., Brum, D. M., & Cassella, R. J. (2015). Determination of metals in soft drinks packed in different materials by ETAAS. *Food Chemistry*, 185, 488–494.
19. Szymczycha-Madeja, A., Welna, M., & Pohl, P. (2013). Determination of elements in energy drinks by ICP OES with minimal sample preparation. *Journal of the Brazilian Chemical Society*, 24(10), 1606–1612.